

AD-A103 913

HOWARD UNIV WASHINGTON D C DEPT OF CHEMISTRY
RAMAN SCATTERING SPECTRA OF TiPO_4 . (U)
JUN 81 S GUHA

F/6 7/4

UNCLASSIFIED

TR-3

N00014-80-C-0305
NL

1-1
A
A



END
DATE
FILMED
O 8
DTIC

REPORT DOCUMENTATION PAGE

LEVEL II

READ INSTRUCTIONS
BEFORE COMPLETING FORM

1. REPORT NUMBER

3

2. GOVT ACCESSION NO.

AD-A103913 P/S

3. RECIPIENT'S CATALOG NUMBER

4. TITLE (and Subtitle)

Raman scattering spectra of TmPO_4 5. TYPE OF REPORT & PERIOD COVERED
Technical Report #3, 1981

6. PERFORMING ORG. REPORT NUMBER

7. AUTHOR(s)

Soumyendu Guha

8. CONTRACT OR GRANT NUMBER(s)

N0014-80-C-0305

9. PERFORMING ORGANIZATION NAME AND ADDRESS

Chemistry Department
Howard University
Washington D.C. 2005910. PROGRAM ELEMENT, PROJECT, TASK
AREA & WORK UNIT NUMBERS

NR 051-733

11. CONTROLLING OFFICE NAME AND ADDRESS

Office of Naval Research
Arlington, Va 22217

12. REPORT DATE

Jun 1981

13. NUMBER OF PAGES

14. MONITORING AGENCY NAME & ADDRESS (if different from Controlling Office)

(14) TIR-3

15. SECURITY CLASS. (of this report)

Unclassified

15a. DECLASSIFICATION/DOWNGRADING
SCHEDULE

16. DISTRIBUTION STATEMENT (of this Report)

Approved for public release : distribution unlimited

17. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from Report)

18. SUPPLEMENTARY NOTES

19. KEY WORDS (Continue on reverse side if necessary and identify by block number)

Raman scattering, TmPO_4

4

20. ABSTRACT (Continue on reverse side if necessary and identify by block number)

Raman scattering spectra of TmPO_4 have been studied between 5° K and room temperature. We have observed four electronic transitions between the crystal field split levels of the Tm^{3+} ion ($^3\text{H}_6$) which is at the D_{2d} site symmetry of the lattice. In addition to electronic transitions, we have observed eleven phonon peaks which we assign to the internal and external vibrations of the PO_4^{3-} ion in the irreducible representation of D_{4h} .

DD FORM 1 JAN 73 1473

EDITION OF 1 NOV 65 IS OBSOLETE
S/N 0102-014-6601

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

81 9 08 61

171660

Jm

AD A103913

DD FORM 1473

Raman Scattering Spectra of $TmPO_4$

Soumyendu Guha
Chemistry Department, Howard University
Washington D.C. 20059

Abstract

Raman Scattering spectra of $TmPO_4$ have been studied between 5° K and room temperature. We have observed four electronic transitions between the crystal field split levels of the Tm^{3+} ion ($3H_6$) which is at the D_{2d} site symmetry of the lattice. In addition to electronic transitions, we have observed eleven phonon peaks which we assign to the internal and external vibrations of the PO_4^{3-} ion in the irreducible representation of D_{4h} .

Like several other rare earth phosphates and vanadates, no phase transition has been observed in this crystal.

Accession For	
NTIS GRA&I	<input checked="checked" type="checkbox"/>
DTIC TAB	<input type="checkbox"/>
Unannounced	<input type="checkbox"/>
Justification	
By	
Distribution/	
Availability Codes	
Dist	Avail and/or Special
A	

DTIC
ELECTE
SEP 8 1981
S D

Introduction

Laser Raman scattering from the electronic states has been found to be a useful method to determine the low lying electronic energy states of rare earth ions in various complexes. In addition, a great deal of information regarding the phase transition in rare earth complexes¹ has also been obtained from electronic and one phonon Raman scattering. This paper reports Raman scattering studies of TmPO_4 crystal between 5° K and room temperature.

The free ion ground electronic state ($^3\text{H}_6$) of the Tm^{3+} ion is split into ten energy levels ($2\text{A}_1 + \text{A}_2 + 2\text{B}_1 + 2\text{B}_2 + 3\text{E}$) in the irreducible representation of D_{2d} which is the site symmetry of rare earth ion in the lattice. Our Raman scattering data at 5° K have revealed four sharp peaks in the low frequency region ($< 300 \text{ cm}^{-1}$) which we identify as the $\text{A}_1^1 \rightarrow \text{E}^1$ (29 cm^{-1}), $\text{A}_1^1 \rightarrow \text{B}_2^1$ (84 cm^{-1}), $\text{A}_1^1 \rightarrow \text{E}^2$ (138 cm^{-1}) and $\text{A}_1^1 \rightarrow \text{E}^3$ (280 cm^{-1}) electronic transitions. The assignments of these levels are in accord with the optical absorption data of Knoll et al.² and ultrasonic sound scattering data of Harley et al.³ Why the transitions between the ground and other excited electronic states were not observed is not understood. It is presumably due to the weak intensities of these peaks or overlapping frequencies with the phonon spectra in the low frequency region ($< 500 \text{ cm}^{-1}$).

We have observed at least twelve sharp phonon peaks at room temperature. The two TmPO_4 molecules per primitive cell in the D_{4h} zircon structure have thirty-six branches in the phonon spectrum. At $q=0$, there are thirteen even parity modes ($2\text{A}_{1g}^{\text{I}} + 2\text{B}_{1g}^{\text{I}} + \text{B}_{2g}^{\text{I}} + 2\text{E}_g^{\text{I}} + \text{A}_{2g}^{\text{Ex}} + 2\text{B}_{1g}^{\text{Ex}} + 3\text{E}_g^{\text{Ex}}$) and all except A_{2g} are Raman active. We have also labelled the internal and external vibrations of the PO_4^{3-} ion with superfixes I and Ex respectively. The positions and intensities of these peaks vary

between room temperature and liquid helium temperature. As will be discussed in the following sections, we attribute it mostly to an anharmonic effect arising due to thermal contraction of the crystal at low temperature.

2. Experimental

Our Raman scattering experiment includes an argon ion laser, a Spex 1400 double monochromator and conventional photon counting detection which is directly connected to a PDP 11 computer. This enabled us to store our data on magnetic tapes and then to plot it with the computer plotter. Most measurements were made with the 5145 Å line at 40 mw power level to avoid local heating of the sample. Slits employed were 80-150-80 which is equivalent to a resolution of $\sim 1 \text{ cm}^{-1}$. The sample was mounted inside a cryogenic dewar and was surrounded by helium exchange gas. Temperatures were measured with gold-iron-chromel thermocouple and were controlled by a PAR 152 cryogenic temperature controller. The dimension of the sample was 3x3x4 mm with the c axis parallel to the longest direction and was obtained from the laboratory of Dr. S.H. Smith at Oxford, England. Four independent components of Raman tensors which transform as the $A_{1g}(\alpha_{xx} + \alpha_{yy}, \alpha_{zz})$, $B_{1g}(\alpha_{xx} - \alpha_{yy})$, $B_{2g}(\alpha_{xy})$, and $E_g(\alpha_{xz}, \alpha_{yz})$ irreducible representations of D_{4h} were measured by suitable combinations of incident and scattered polarisations. The conventional 90° scattering geometry were employed to observe the Raman data.

3. Experimental Results and Discussion

3.1. Electronic Raman Spectra

Strong electronic peaks appear between 0 and 300 cm^{-1} below 77° K . The electronic peak at 84 cm^{-1} (Figure 1a) in the B_2 spectrum is as

strong (6300 counts/sec or 5 mv dc) as some strong phonon peaks. We identify this peak as the $A_1^1 \rightarrow B_2^1$ electronic transition. The choice of the ground state singlet as A_1 is purely arbitrary. At 77°K , a peak at 108 cm^{-1} also appears in the B_2 polarisation (Fig. 2) which we identify as the $E^1 \rightarrow E^2$ electronic transition. This 108 cm^{-1} peak also appears in the $A_1 + B_1$ polarisation at 77°K which justifies our assertion to this peak to the $E \rightarrow E$ transition.

We have also observed three electronic peaks at 5°K in the E polarisation geometry which we identify as the $A_1^1 \rightarrow E$ (29 cm^{-1}), $A_1^1 \rightarrow E^2$ (138 cm^{-1}) and $A_1^1 \rightarrow E^3$ (280 cm^{-1}). The intensities of these peaks are either comparable or an order of magnitude less than the weaker phonon lines (~ 200 counts/sec or 1 mv dc). A considerable amount of leakage from all polarisation geometries has been observed. We have associated the observed peaks to different polarisation by carefully monitoring the intensities of these peaks. For example, the strong peak at 84 cm^{-1} in the B_2 polarisation (Figure 1) appears as a weak line in the E polarisation and therefore, we regard this peak as a leakage to the E spectrum from the B_2 spectrum.

According to group theory, there are six singlets ($2A_1 + 2B_1 + 2B_2$) in the ground state manifold of the Tm^{3+} ion at the D_{2d} site symmetry. Electronic transitions between these energy levels are all Raman allowed; except the $A_1 \rightarrow B_2$ transition, we were unable to observe any other electronic transitions among these singlets. Crystal field calculations on Tm^{3+} ; $\text{YVO}_4^{(2)}$ indicate that the energy levels arising due to the crystal field splitting of the ground state of Tm^{3+} extend over to 365 cm^{-1} . Although the crystal fields at the rare earth

site in the phosphate differs from the vanadates, it is unlikely that the crystal field parameters would change the energy levels significantly in these two systems. We have arrived at this conclusion by comparing the energy level schemes in $\text{Eu}^{3+}:\text{YVO}_4, \text{YPO}_4^{(4)}$ and $\text{Er}^{3+}:\text{YVO}_4, \text{YPO}_4^{(5)}$. For both of these rare earth ions, the energy levels of the ground state manifold do not differ substantially from the phosphates to vanadates as regards to the spread of the energy states i.e. the separation between the ground and final excited state of the ground state manifold. The failure to observe the other singlets is probably due to the weak intensities of these electronic transitions. Another possible explanation is that the frequencies of these singlets lie close to the low lying phonon frequencies which are probably masking them. The observed energy levels are shown in Table 1.

3.2. Phonon Spectra

The Raman active phonon modes can be classified as the internal vibrations of the phosphate ion and external modes in which the phosphate moves as a unit relative to rare earth. The identification of the observed phonon peaks (Figure 2) is achieved by carefully monitoring the intensities of these peaks in different polarisations. As can be seen in Figure 2, a leakage (labelled as L) from one polarisation geometry to the other has been a considerable problem in assigning the observed phonon frequencies to a particular polarisation geometry. The Table 2 in reference 6 has been a considerable help to us. We have also quoted the corresponding phonon frequencies in $\text{DyPO}_4^{(6)}$ in Table 2 for comparison. The symmetric stretching mode of free phosphate ion occurs around 980 cm^{-1} . This is also the most intense peak. We assign the 1057 cm^{-1} peak to the symmetric stretching mode (A_{1g}) although it is not clear why it has

been shifted to higher frequency in TmPO_4 . Moreover, we have observed a change in frequencies for all external and internal modes as compared to DyPO_4 . It is possible that the particular crystal we have used may have internal strains; alternatively a strong electron lattice coupling may be responsible for such a large shift in the observed frequencies in these two systems. Another anomalous trend that we have observed is the temperature dependency of the phonon frequencies. Usually, a decreased temperature results in lattice contraction which leads to increased force constants and higher frequencies. It has been reported⁽³⁾ that the elastic constant c_{66} in TmPO_4 exhibits a marked softening at 20°K. It is probable that a combined effect of the anharmonicity due to thermal contraction and softening of the force constant due to electron lattice coupling is responsible for the anomalous trend of phonon frequencies as a function of temperature.

Finally, we would like to mention that we have not observed any phase transition in this crystal till 5°K. We have carefully investigated the phonon and electronic peaks in the vicinity of this temperature and have not noticed any splitting or significant change in the spectrum, although both phonon and electronic peak started becoming sharper at lower temperature.

4. Conclusion

We have observed four sharp electronic transitions in the low frequency Raman spectra at 5.5°K which we have identified as the transitions between the ground state manifold of the Tm^{3+} ion. Although the group theory predicts ten electronic energy levels, we were able to observe only five of them. In addition, we have also observed eleven Raman active one phonon scattering modes.

Acknowledgments

The author expresses his gratitude to Prof. L. L. Chase of Indiana University for providing the laboratory facilities where most of the low temperature (5° K) work was done. Thanks are due to Dr. R. Prater of Indiana University for computational help with the FDP 11 computer. The author also wishes to thank Prof. R. S. Katiyar of University of Campinas, Brasil for providing the sample. This work is partially by an ONR laser chemistry grant at Howard University.

References

1. Gehring G A and Gehring K A, Rep. Prog. Phys., 38, 1, 1975.
2. Knoll K D, Phys. Stat. Solidi (b), 45, 553, 1971.
3. Harley T and Manning D I, J. Phys. C: Solid State Phys., 11, L633, 1978.
4. Brecher C, Samelson H, Lempicki A, Riley R, Peters T, Phys. Rev., 155, 178, 1967; J. Chem. Phys., 40, 3303, 1968.
5. Kuse D, Z. Phys., 203, 49, 1967.
6. Elliot R J, Harley R T, Hayes W and Smith S R P, Proc. R. Soc. Lond. A, 328, 217, 1972.

Table 1. Frequencies in cm^{-1} of the observed electronic transitions between the energy levels of the ground state manifold of the Tm^{3+} ion.

A_1^1	E^1	B_2^1	E^2	E^3
0	29	84	138	280

Table 2. Frequencies in cm^{-1} of the Raman active phonon modes of TmPO_4 (Accuracy of measurements is $\pm 2 \text{ cm}^{-1}$). For comparison, the corresponding frequencies of internal and external modes of DyPO_4 in reference 6 are also listed.

		A_{1g}	E_g	B_{1g}	B_{1g}	E_g	A_{1g}	B_{2g}					
									B_{1g}	E_g	E_g	B_{1g}	E_g
TmPO_4	300 K	1057	1086	1123	678	597	499	336	-	308	188	141	134
	77 K	1055	1077	1125	683	593	498	334	-	312	185	139	132
	5 K	1051	1075	1117	678	591	497	336	-	309	188	135	133
$\text{DyPO}_4^{(6)}$	300 K	997	1017	1053	653	576	485	330	-	294	184	140	130
	77 K	998	1019	1057	655	577	488	330	-	297	186	140	132
	4.5 K	999	1021	1060	656	578	491	328	-	296	-	138	132

Figure Captions

Figure 1. Low frequency spectra at 5°K . (a) The B_2 (xy) spectrum shows a strong peak at 84 cm^{-1} which is identified as the $A_1^1 \rightarrow B_2$ electronic transition. (b) The E (xz) spectrum shows three electronic transitions which are identified as the $A_1^1 \rightarrow E^1$ (20 cm^{-1}), $A_1^1 \rightarrow E^2$ (138 cm^{-1}) and $A_1^1 \rightarrow E^3$ (280 cm^{-1}) transitions. The peak at 84 cm^{-1} is a leakage from the B_2 polarisation, while the peaks at 133 cm^{-1} and 308 cm^{-1} are phonon lines.

Figure 2. One-phonon Raman spectra at 77°K . Different polarisation geometry spectra are shown in (a), (b), (c) and (d). Here L indicates the leakage from one geometry to the other, while e indicates the electronic Raman transitions which are described in Figure 1. The most intense lines are observed in A_{1g} (c) and A_{1g}, B_{1g} (d) spectra. The intensity of the 1057 cm^{-1} is about 6300 counts/sec, while that of 1125 cm^{-1} peak is 5000 counts/sec.

//

INTENSITY (ARBITRARY UNITS)

(a)

x 1

$X(YX)Y (B_2)$

+

(b)

$X(YZ)Y (E)$

x 10

29

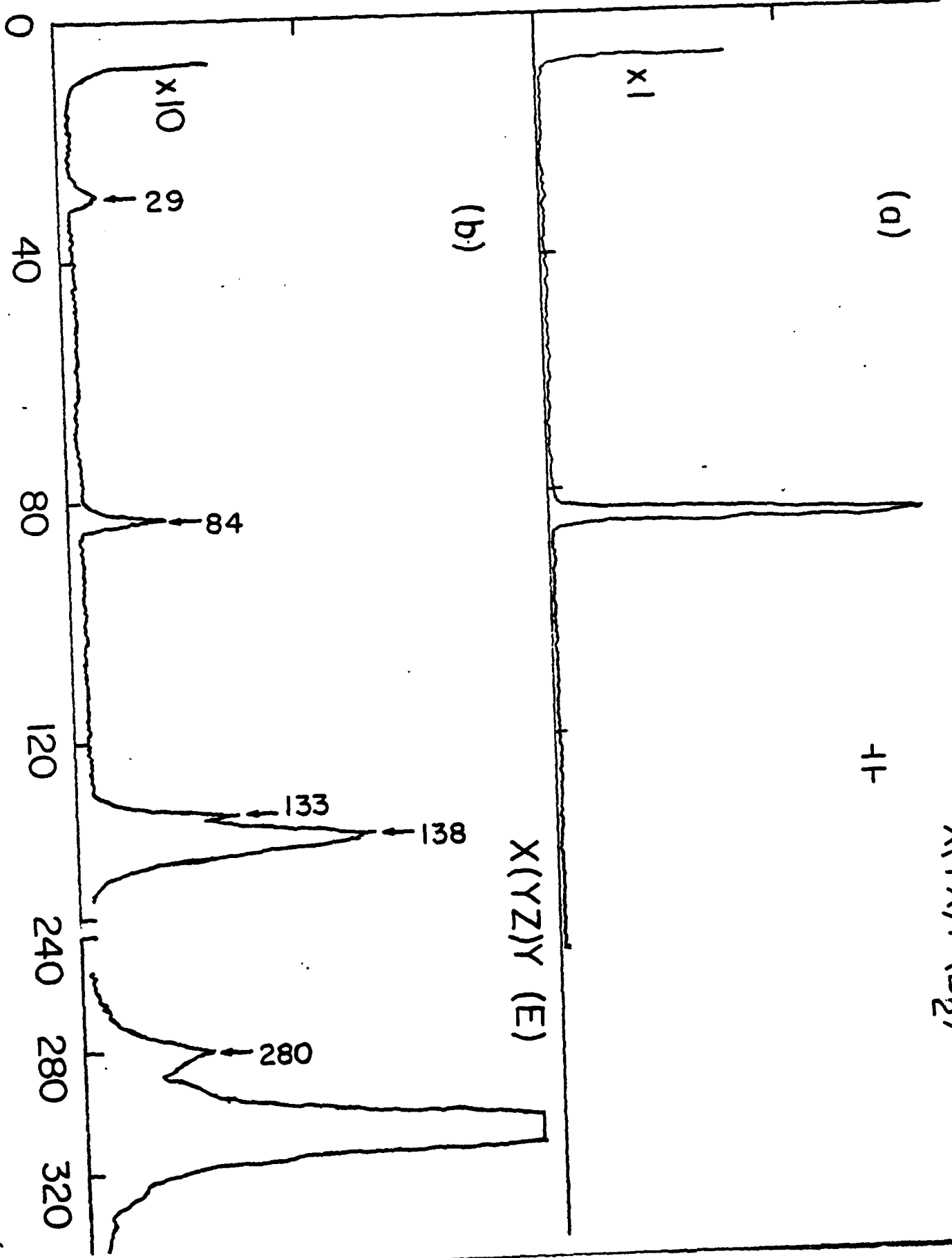
84

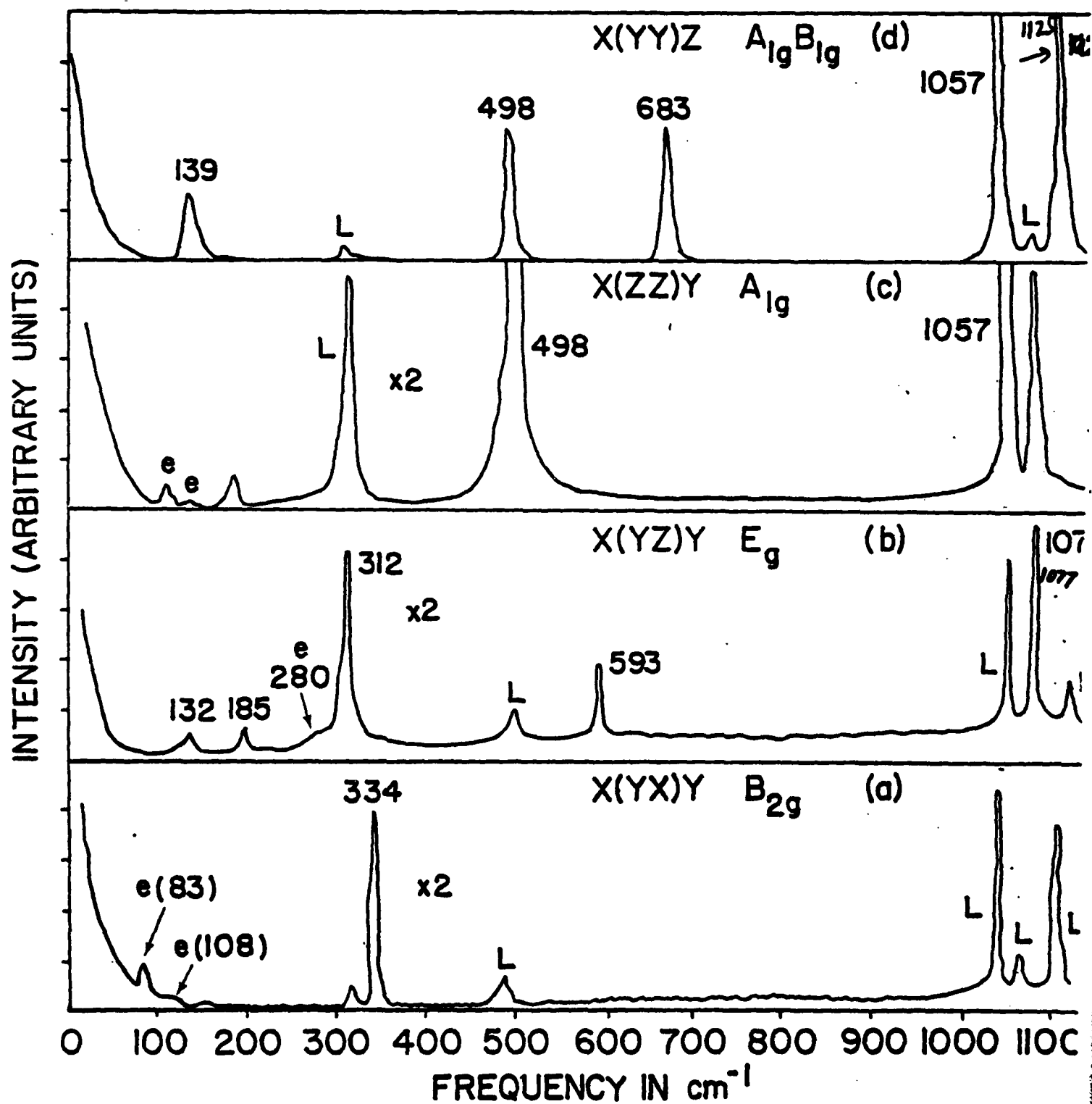
133

138

280

FREQUENCY (cm^{-1})





TECHNICAL REPORT DISTRIBUTION LIST, GEN

	<u>No.</u> <u>Copies</u>		<u>No.</u> <u>Copies</u>
Office of Naval Research Attn: Code 472 800 North Quincy Street Arlington, Virginia 22217	2	U.S. Army Research Office Attn: CRD-AA-IP P.O. Box 1211 Research Triangle Park, N.C. 27709	1
ONR Branch Office Attn: Dr. George Sandoz 536 S. Clark Street Chicago, Illinois 60605	1	Naval Ocean Systems Center Attn: Mr. Joe McCartney San Diego, California 92152	1
ONR Area Office Attn: Scientific Dept. 715 Broadway New York, New York 10003	1	Naval Weapons Center Attn: Dr. A. B. Amster, Chemistry Division China Lake, California 93555	1
ONR Western Regional Office 1030 East Green Street Pasadena, California 91106	1	Naval Civil Engineering Laboratory Attn: Dr. R. W. Drisko Port Hueneme, California 93401	1
ONR Eastern/Central Regional Office Attn: Dr. L. H. Peebles Building 114, Section D 665 Summer Street Boston, Massachusetts 02210	1	Department of Physics & Chemistry Naval Postgraduate School Monterey, California 93940	1
Director, Naval Research Laboratory Attn: Code 6100 Washington, D.C. 20390	1	Dr. A. L. Slafkosky Scientific Advisor Commandant of the Marine Corps (Code RD-1) Washington, D.C. 20380	1
The Assistant Secretary of the Navy (RE&S) Department of the Navy Room 4E736, Pentagon Washington, D.C. 20350	1	Office of Naval Research Attn: Dr. Richard S. Miller 800 N. Quincy Street Arlington, Virginia 22217	1
Commander, Naval Air Systems Command Attn: Code 310C (H. Rosenwasser) Department of the Navy Washington, D.C. 20360	1	Naval Ship Research and Development Center Attn: Dr. G. Bosmajian, Applied Chemistry Division Annapolis, Maryland 21401	1
Defense Technical Information Center Building 5, Cameron Station Alexandria, Virginia 22314	12	Naval Ocean Systems Center Attn: Dr. S. Yamamoto, Marine Sciences Division San Diego, California 91232	1
Dr. Fred Saalfeld Chemistry Division, Code 6100 Naval Research Laboratory Washington, D.C. 20375	1	Mr. John Boyle Materials Branch Naval Ship Engineering Center Philadelphia, Pennsylvania 19112	1

TECHNICAL REPORT DISTRIBUTION LIST, 051A

	<u>No. Copies</u>		<u>No. Copies</u>
Dr. M.A. El-Sayed Department of Chemistry University of California, Los Angeles Los Angeles, California 90024	1	Dr. M. Raubut Chemical Research Division American Cyanamid Company Bound Brook, New Jersey 08805	1
Dr. E. R. Bernstein Department of Chemistry Colorado State University Fort Collins, Colorado 80521	1	Dr. J. I. Zink Department of Chemistry University of California, Los Angeles Los Angeles, California 90024	1
Dr. C. A. Heller Naval Weapons Center Code 6059 China Lake, California 93555	1	Dr. D. Haarer IME San Jose Research Center 5600 Cottle Road San Jose, California 95143	1
Dr. J. R. MacDonald Chemistry Division Naval Research Laboratory Code 6110 Washington, D.C. 20375	1	Dr. John Cooper Code 6130 Naval Research Laboratory Washington, D.C. 20375	1
Dr. G. B. Schuster Chemistry Department University of Illinois Urbana, Illinois 61801	1	Dr. William M. Jackson Department of Chemistry Howard University Washington, DC 20059	1
Dr. A. Adamson Department of Chemistry University of Southern California Los Angeles, California 90007	1	Dr. George E. Walraffen Department of Chemistry Howard University Washington, DC 20059	1
Dr. M. S. Wrighton Department of Chemistry Massachusetts Institute of Technology Cambridge, Massachusetts 02139	1		

No.
Copies

Dr. Rudolph J. Marcus
Office of Naval Research
Scientific Liaison Group
American Embassy
APO San Francisco 96503

1

Mr. James Kelley
DTNSRDC Code 2803
Annapolis, Maryland 21402

1